立中央大學八十六學年度碩士班研究生入學試題卷

化學研究所 不分組

無機化學 共 頁 第 / 頁

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- The Pk value for arsenious acid, H₃AsO₃, is 9.2 and that for hypophosphorus acid, H₃PO₂, is ~1. What structures are consistent with these values ? (6%)
- 2. The 18e rule is a way to help us decide whether a give d-block transition metal organometallic complex is likely to be stable. please give three complexes with 18e, and also three complexes without 18e but stable. (12%)
- 3. In the presence of a base; OH', hydrolysis of [Co(NH₃)₅Cl] ²⁺ gives Cl displacement product; [Co(NH₃)₅OH] ²⁺. Please propose a mechanism for this substitution reaction. (10%)
- 4. If substition reaction of octahedral complexes proceeds by dissociative mechanism via a square pyramid intermediate, stereochemistry of reaction products is retented. (i.e. cis gives 100% cis or trans gives 100% trans). If reaction proceeds by dissociative mechanism but via TBP intermediate, please discuss the stereochemistry of the products. (10%)
- 5. Rh(PPh₃)₃Cl complex is an efficient catalyst as Wilkinson's catalyst for olefin hydrogenation. Please propose this catalytic cycle starting from CH₂=CHR. (12%)
- 6. What is ligand field theory ? What kind of properties can be obtained from this theory ? (10%).
- 7. What are the differences between molecular solids and solid state molecules ? (5%).
- 8. Why is DH_{vap} for water much greater than DH_{fus}? What does this reveal concerning changes in intermolecular forces in going from solid to liquid to vapor ? (5%)
- 9. A Lewis structure obeying the octet rule can be drawn for O₂ as follows:

$$\dot{o} = \dot{o}$$

Use the molecular orbital energy-level diagram for O2 to show that this Lewis structure for O2 corresponds to an excited state. (10%).

- 10. Using the p orbitals for an example, distinguish between the angular part of the probability function, the radial part of the probability function, and a probability contour. Draw simple sketches to illustrate. How could each of these be affected by a change in the principal quantum number, n? (10%)
- 11. The Cope rearrangement is as follows:



carry out the appropriate symmetry analysis to show that it is thermally allowed. (10%).