學年度碩士班研究生入學試題卷

科目:

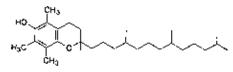
- (一) 是非題 (以下敘述正確者, 在答案紙盤 O; 敘述錯誤者數 X; 每題 3 分, 答對得 3 分; 答錯倒扣 2 分; 不答不計分,本大題共 60 分)
- (1) A correct name for the compound shown below is isopentyl methyl ketone.

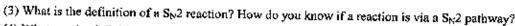
- (2) The order of decreasing reactivity of aldehydes, esters, and ketones towards the organozine compounds used in the Reformatsky reaction is Aldehydes > Esters > Ketones.
- (3) The final product of the following reaction is ethyl 4-chlorobenzoate.

КМлО₄ H* SOCi₂ _ 4-chlorotoluene

- (4) Cl₂CCOOH has a smaller value of pKa than CH₂COOH.
- (5) (CH₃)₃CBr cannot be used effectively in a Gabriel amine synthesis.
- (6) Neopentyl isobutyrate has the IUPAC name: 2,2-dimethylpropyl 2-methylpropanoate.
- (7) Removal of a hydride ion from cyclopentadiene produces an aromatic cation.
- (8) In a Diels-Alder reaction, the adduct formed most rapidly is generally the exo product.
- (9) A thermodynamically-controlled reaction will yield predominantly the more/most stable product.
- (10) CrO₃ in H₂SO₄ can serve as the basis for a simple chemical test to distinguish between CH₃CH∞CHBr and $CH_2=CHCH_2Br.$

- (13) The activation energy is zero in the chain-terminating step of a radical chain reaction.
- (14) The reaction of Oxymercuration-demercuration of alkenes follows the Markovnikov's rule.
- (15) HOCH2CH2CH2CH2CH2OH has a higher boiling point than CH3OCH2CH2CH2OCH3 because of larger dispersion forces.
- (16) Molecules containing a stereocenter must be chiral.
- (17) The stronger the acid, the more positive the value of ΔG° for the dissociation.
- (18) The dipole moments of CO₂ and CCl₄ are zero.
- (19) The J^3 coupling constant of H-C=C-H in proton NMR is trans isomer < eis isomer.
- (20) The nitro group has a pair of strong IR band at 1650 and 1450 cm⁻¹.
- (二) 回答下列問題: (每題 10 分, 共 40 分)
- (1) Propose a method for the conversion of stilbene (Ph-C=C-Ph) to diphenylacetylene (Ph-C=C-Ph).
- (2) Vitamin E (a-tocopherol)(structure is shown below) is capable of acting as a radical trap, and one of the important roles that Vitamin E plays in the body may be in inhibiting radical reactions that could cause cell damage. Propose a mechanism to account for the antioxidation behavior of Vitamin E.





(4) What are the definitions of secondary testions of

