

# 國立中央大學九十學年度碩士班研究生入學試題卷

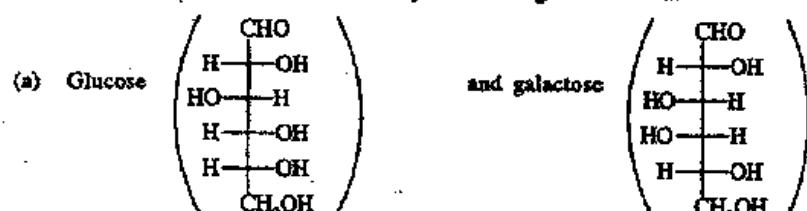
所別： 生命科學系 不分組 科目： 有機化學 共 2 頁 第 1 頁

1. (15 pts)

Write the structural formula of the principal ionic species present in aqueous solution at pH 2, 7, and 12 of isoleucine (2-amino-3-methylpentanoic acid).

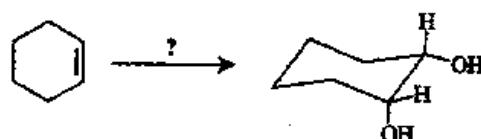
2. (10 pts)

Outline chemical tests that would allow you to distinguish between:



3. (5 pts)

Which reaction sequence converts cyclohexene to cis-1,2-cyclohexanediol? That is,



- (a) Cold, dilute, aqueous  $\text{KMnO}_4$ ,  $\text{OH}^-$     (b) (1)  $\text{O}_3$     (2)  $\text{Zn}/\text{HOAc}$   
 (c) (1)  $\text{OsO}_4$     (2)  $\text{NaHSO}_3$     (d) (1)  $\text{RCOOH}$     (2)  $\text{H}_3\text{O}^+/\text{H}_2\text{O}$   
 (e) More than one of these

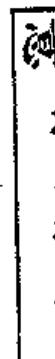
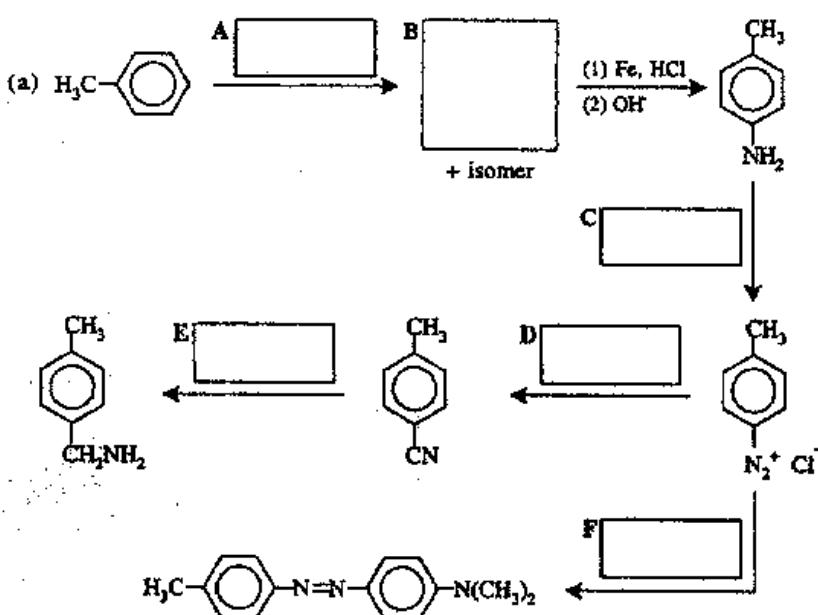
4. (5 pts)

Which of the following sequences leads to the best synthesis of the compound  $\text{CH}_3\text{CH}_2\text{C}\equiv\text{CH}$ ? (Assume that the quantities of reagents are sufficient to carry out the desired reaction.)

- (a)  $\text{CH}_3\text{CH}_2\text{CH}=\text{CH}_2 \xrightarrow{\text{Br}_2} \xrightarrow{\substack{\text{NaOH} \\ \text{H}_2\text{O}}} \text{CH}_3\text{CH}_2\text{CH}_2\text{CHBr}_2$   
 (b)  $\text{CH}_3\text{CH}_2\text{CH}=\text{CH}_2 \xrightarrow{\text{Br}_2} \xrightarrow{\text{NaNH}_2} \text{CH}_3\text{CH}_2\text{CH}_2\text{CHBr}_2$   
 (c)  $\text{CH}_3\text{CH}_2\text{CH}_2\text{CHBr}_2 \xrightarrow{\text{H}_2\text{SO}_4}$   
 (d)  $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3 \xrightarrow[\text{light}]{\text{Br}_2} \xrightarrow{\substack{\text{NaNH}_2 \\ \text{Zn}, \text{HOAc}}} \text{CH}_3\text{CH}_2\text{CH}=\text{CH}_2$   
 (e)  $\text{CH}_3\text{CH}_2\text{CH}=\text{CH}_2 \xrightarrow{\text{O}_3} \xrightarrow{\text{Zn}, \text{HOAc}}$

5. (24 pts)

Complete the following syntheses:

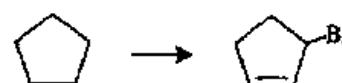


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6. (5 pts)

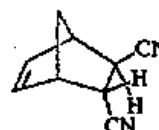
Which reagent(s) could be used to carry out the following reaction?



- (a) NBS/CCl<sub>4</sub> (NBS = )
- (b) NBS/CCl<sub>4</sub>, then Br<sub>2</sub>/hν
- (c) Br<sub>2</sub>/hν, then (CH<sub>3</sub>)<sub>3</sub>COK/(CH<sub>3</sub>)<sub>3</sub>COH, then NBS/CCl<sub>4</sub>
- (d) (CH<sub>3</sub>)<sub>3</sub>COK/(CH<sub>3</sub>)<sub>3</sub>COH, then NBS/CCl<sub>4</sub>

7. (5 pts)

Which diene and dienophile could be used to synthesize the following compound?



- (a) +
- (b) +
- (c) +
- (d) +
- (e) +

8. (15 pts)

Select the reagent from the list below that could be the basis of a simple chemical test that would distinguish between each of the following:

- (a) C<sub>6</sub>H<sub>5</sub>CNH<sub>2</sub> and C<sub>6</sub>H<sub>5</sub>CH<sub>2</sub>NH<sub>2</sub>
- (b) C<sub>6</sub>H<sub>5</sub>CH<sub>2</sub>NH<sub>2</sub> and C<sub>6</sub>H<sub>5</sub>NHCH<sub>3</sub>
- (c) and

1. Cold dilute NaHCO<sub>3</sub>,
2. Cold dilute HCl
3. NaNO<sub>2</sub>, HCl, 5°C, then 2-naphthol
4. C<sub>6</sub>H<sub>5</sub>SO<sub>2</sub>Cl, OH<sup>-</sup>, then HCl
5. Cold dilute NaOH

9. (16 pts)

Complete the following synthesis:

