

國立中央大學 110 學年度碩士班考試入學試題

所別：化學工程與材料工程學系 碩士班 甲組(一般生)

共 4 頁 第 1 頁

科目：輸送現象與單元操作

本科考試可使用計算器，廠牌、功能不拘

* 請在答案卷(卡)內作答

※ 計算題需計算過程，無計算過程者不予計分

1. (20%) Please explain each of the terms in the following (2 pts each).

(a) Reynold number, (b) molecular momentum flux, (c) Prandtl mixing length, (d) velocity potential, (e) Reynolds stress, (f) form drag, (g) fanning friction factor, (h) eddy diffusivity, (i) divergence of velocity, (j) creeping flow.

2. (8%) Introduce the following dimensionless terms into the Navier-Stokes equation:

$u^* = u/u_\infty$, dimensionless velocity

$P^* = P/\rho u_\infty^2$, dimensionless pressure

$t^* = t u_\infty / L$, dimensionless time

$x^* = x / L$, dimensionless distance

The operator ∇ may then written ∇^*/L . Show that the Navier-Stokes equation becomes

$$\frac{Du^*}{Dt^*} = \frac{gL}{u_\infty^2} - \nabla^* P^* + \frac{1}{Re} \nabla^{*2} u^*$$

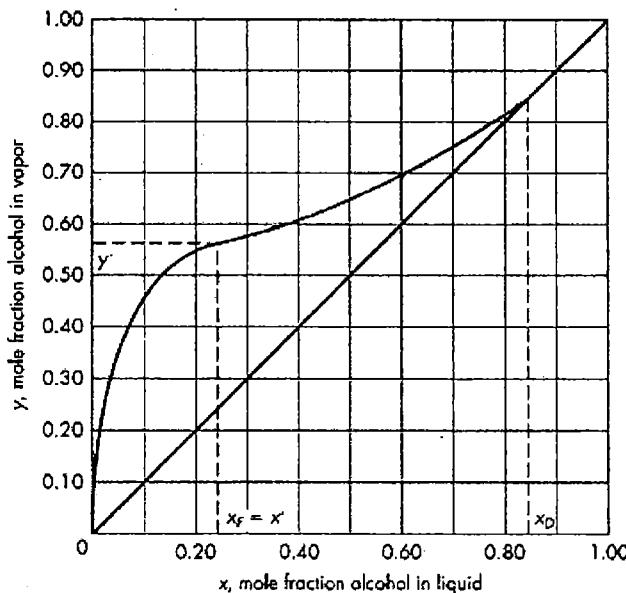
3. (7%) A packed bed is composed of cylinders having a diameter $D=0.02$ m and a length $h=D$. The bulk density of the overall packed bed is 962 kg/m^3 and the density of the solid cylinders is 1600 kg/m^3 .

(a) (2 pts) Calculate the void fraction ε .

(b) (3 pts) Calculate the effective diameter D_p of the particles.

(c) (2 pts) Calculate the ratio of total surface area in the bed to total volume of bed in m^{-1}

4. (10%) Please recreate the diagram provided below in your answer sheet, and draw the "minimum reflux ratio" and "minimum operating ratio" operating lines. Clearly label the two lines.



注意：背面有試題

國立中央大學 110 學年度碩士班考試入學試題

所別：化學工程與材料工程學系 碩士班 甲組(一般生)

共 4 頁 第 2 頁

科目：輸送現象與單元操作

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5. (25%) A pipeline transporting cold water is surrounded by stagnant air that is both hot and humid, and moisture continually diffuses to the cold surface of the pipe to form condensation. The condensed water forms a liquid film around the pipe, which drops off. At a distance of 10 cm from the exterior of the pipe, the relative humidity is constant, while close to the pipe the moisture content approaches the vapor pressure of water evaluated at the temperature of the pipe.

- Which coordinate system best describes this transfer process? List three major assumptions. (5 pts)
- Provide the simplified form of the general differential equation for mass transfer in terms of the flux of water vapor, N_A . (5 pts)
- What is the simplified form of Fick's flux equation for water vapor, N_A ? (5 pts)
- Provide the simplified form of the general differential equation for mass transfer in terms of the concentration of water vapor, C_A ? (10 pts)

6. (30%) A team of engineers for a defense contractor are commissioned by the naval force of a global power to build a nuclear submarine. The specs for the submarine, as demanded by the navy in order to designate the submarine as the Los Angeles Class, is as follows,

- Weight: 7,000 tones (1 tone = 1,000 kg)
- Top speed when submerged: 20 knots (~ 37 km/hr)
- Acceleration from rest to top speed when submerged: 1 minute

It is also estimated that all the friction experienced by a submarine submerged in sea water is 3.86×10^6 N. Moreover, such a submarine is conventionally propelled by a steam-driven turbine, with the high-pressure steam generated from a boiler heated by the nuclear reaction. Now, you are assigned the most critical part of the project – design the heat exchange system for the power generation system of the submarine – and need to determine the operation condition of the heat exchanger. Before that can be done, several other operation parameters have to be known in advance. (1) In the steady-state operation, steam leaving the turbine is condensed and recycled back to the boiler. For the sake of efficacy, steam from the outlet of the turbine is set to 1 bar and 100°C. Given the steam flow rate of 20 kg/s, determine the minimal temperature and pressure of steam entering the turbine (assuming entropy generation $S_{gen} = 0$). (4 pts) (2) Given that compressed water of 100°C and 10 MPa is steadily supplied to the boiler, determine the operation temperature of the boiler and the heat generation rate which must be achieved by the nuclear reaction to maintain the steady and saturated state of steam at the pressure determined in (1). (4 pts) (3) A temperature discrepancy thus exists between steam leaving the boiler and the one entering the turbine. Making up the discrepancy stands the reason for deploying a heat exchanger in between the boiler and the turbine. Determine the overall heat transfer rate of the heat exchanger. (4 pts) (4) You decide to employ a counterflow, concentric tube heat exchanger for the purpose. Given the condition that the cooling fluid – water – is at 10°C and the diameters of the outer annulus and inner tube are to be 45 cm and 25 cm, respectively, determine the respective convection heat transfer coefficients for cooling water and steam. (9 pts) Explain the physical meanings of Reynolds and Nusselt numbers, which are relevant for addressing the question. (2 pts) (5) With the results of (4), determine the overall heat transfer coefficient and thereby the length of the heat exchanger to be built. (7 pts) (Important: Make every necessary assumption and give your reasoning. This will be taken into

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共 4 頁 第 3 頁

科目：輸送現象與單元操作

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consideration when your answers are graded.)

Some useful information

- $\vec{F} = m\vec{a}$; $W = \vec{F} \cdot \vec{s}$
- $Nu_D = 0.023 Re_D^{4/5} Pr^n$; $Nu_D = 0.027 Re_D^{4/5} Pr^{1/3} (\mu/\mu_s)^{0.14}$

Superheated Vapor (Continued)

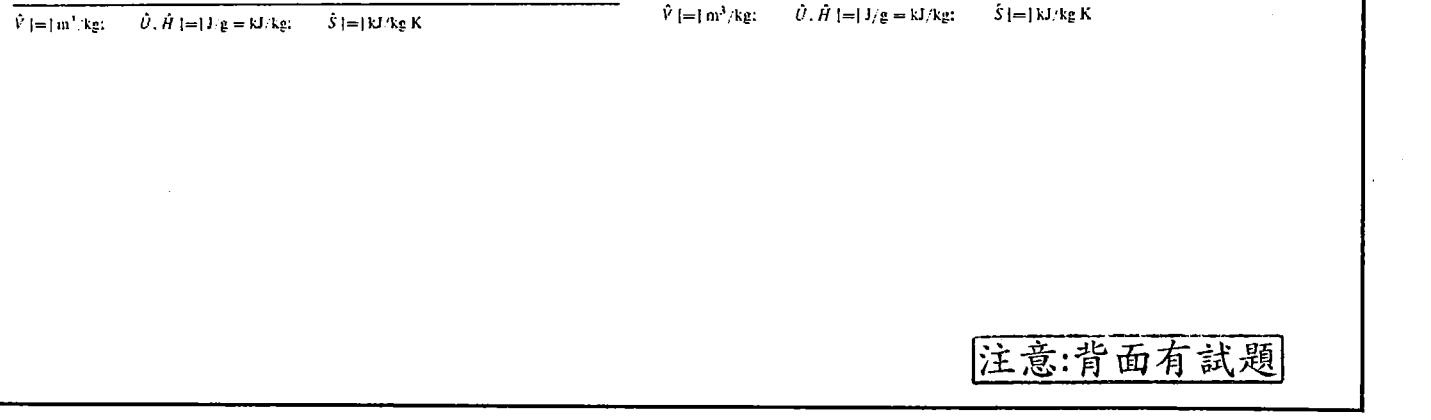
T (°C)	$P = 4.0 \text{ MPa}$ (250.40)			$P = 4.5 \text{ MPa}$ (257.49)		
	\hat{V}	\hat{U}	\hat{H}	\hat{S}	\hat{V}	\hat{U}
Sat.	0.049 78	2602.3	2801.4	6.0701	0.044 06	2600.1
275	0.054 57	2667.9	2886.2	6.2285	0.047 30	2650.3
300	0.058 84	2725.3	2960.7	6.3615	0.051 35	2712.0
350	0.066 45	2826.7	3092.5	6.5821	0.058 40	2817.8
400	0.073 41	2919.9	3213.6	6.7690	0.064 75	2913.3
450	0.080 02	3010.2	3320.3	6.9363	0.070 74	3005.0
500	0.086 43	3099.5	3445.3	7.0901	0.076 51	3095.3
600	0.098 85	3279.1	3674.4	7.3688	0.087 65	3276.0
700	0.110 95	3462.1	3905.9	7.6198	0.098 47	3459.9
800	0.122 87	3650.0	4141.5	7.8502	0.109 11	3648.3
900	0.134 69	3845.6	4382.3	8.0647	0.119 65	3842.2
1000	0.146 45	4042.9	4628.7	8.2662	0.130 13	4041.6
1100	0.158 17	4248.0	4880.6	8.4567	0.140 56	4246.8
1200	0.169 87	4458.6	5138.1	8.6376	0.150 98	4457.5
1300	0.181 56	4674.3	5400.5	8.8100	0.161 39	4673.1
$P = 6.0 \text{ MPa}$ (275.64)						
$P = 7.0 \text{ MPa}$ (285.88)						
Sat.	0.032 44	2589.7	2784.3	5.8892	0.027 37	2580.5
300	0.036 16	2667.2	2884.2	6.0674	0.029 47	2632.2
350	0.042 23	2789.6	3043.0	6.3335	0.035 24	2769.4
400	0.047 39	2892.9	3177.2	6.5408	0.039 93	2878.6
450	0.052 14	2988.9	3301.8	6.7193	0.044 16	2978.0
500	0.056 65	3082.2	3422.2	6.8803	0.048 14	3073.4
550	0.061 01	3174.6	3540.6	7.0288	0.051 95	3167.2
600	0.065 25	3266.9	3658.4	7.1677	0.055 65	3260.7
700	0.073 52	3453.1	3894.2	7.4234	0.062 83	3448.5
800	0.081 60	3643.1	4132.7	7.6506	0.069 81	3639.5
900	0.089 58	3837.8	4375.3	7.8727	0.076 69	3835.0
1000	0.097 49	4037.8	4622.7	8.0751	0.083 50	4035.3
1100	0.105 36	4243.3	4875.4	8.2661	0.090 27	4240.9
1200	0.113 21	4454.0	5133.3	8.4474	0.097 03	4451.7
1300	0.121 06	4669.5	5396.0	8.6199	0.103 77	4667.3
$P = 9.0 \text{ MPa}$ (303.40)						
$P = 10.0 \text{ MPa}$ (311.06)						
Sat.	0.020 48	2557.8	2742.1	5.6772	0.018 026	2544.4
325	0.023 27	2646.6	2856.0	5.8712	0.019 861	2610.4
350	0.025 80	2724.4	2956.6	6.0361	0.022 42	2699.2
400	0.029 93	2848.4	3117.8	6.2854	0.026 41	2824.2
450	0.033 50	2955.2	3256.6	6.4844	0.029 75	2943.4
500	0.036 77	3055.2	3386.1	6.6576	0.032 79	3045.8
550	0.039 87	3152.9	3511.0	6.8142	0.035 64	3144.6
600	0.042 85	3248.1	3633.7	6.9589	0.038 37	3241.7
650	0.046 74	3343.6	3755.3	7.0943	0.041 01	3338.2
700	0.048 57	3439.3	3876.5	7.2221	0.043 58	3434.7
800	0.054 09	3632.5	4119.3	7.4596	0.048 59	3628.9
900	0.059 50	3829.2	4364.8	7.6783	0.053 49	3826.3
1000	0.064 85	4030.3	4614.0	7.8821	0.058 32	4027.8
1100	0.070 16	4236.3	4867.7	8.0740	0.063 12	4234.0
1200	0.075 44	4447.2	5126.2	8.2556	0.067 89	4444.9
1300	0.080 72	4662.7	5389.2	8.4284	0.072 65	4640.5

Compressed Liquid

T (°C)	$P = 5 \text{ MPa}$ (263.99)			$P = 10 \text{ MPa}$ (311.06)		
	\hat{V}	\hat{U}	\hat{H}	\hat{S}	\hat{V}	\hat{U}
Sat.	0.001 285 9	1147.8	1154.2	2.9202	0.001 452 4	1393.0
0	0.000 997 7	0.04	5.04	0.0001 000 995 2	0.09	10.04
20	0.000 999 5	83.65	88.65	0.000 997 2	83.36	93.33
40	0.001 005 6	166.95	171.97	0.001 003 4	166.35	176.38
60	0.001 014 9	250.23	255.30	0.001 012 7	249.36	259.49
80	0.001 026 8	333.72	338.85	0.001 024 5	332.59	342.83
100	0.001 041 0	417.52	422.72	0.001 038 5	416.12	426.50
120	0.001 057 6	501.80	507.09	0.001 054 9	500.08	510.64
140	0.001 076 8	586.76	592.15	0.001 073 7	584.68	595.43
160	0.001 098 8	672.62	678.12	0.001 095 3	670.13	681.08
180	0.001 124 0	759.63	765.25	0.001 119 9	756.65	767.84
200	0.001 153 0	848.1	853.9	0.001 148 0	844.5	856.0
220	0.001 186 6	938.4	944.4	0.001 180 5	934.1	945.9
240	0.001 226 4	1031.4	1037.5	0.001 218 7	1026.0	1038.1
260	0.001 274 9	1127.9	1134.3	0.001 264 5	1121.1	1133.7
280					0.001 321 6	1220.9
300					0.001 397 2	1328.4
320						1342.3
340						1349.6
$P = 20 \text{ MPa}$ (365.81)						
Sat.	0.002 036	1785.6	1826.3	4.0139		
0	0.000 990 4	0.19	20.01	0.0004	0.000 985 6	0.25
20	0.000 992 8	82.77	102.62	0.2923	0.000 988 6	82.17
40	0.000 999 2	165.17	185.16	0.5646	0.000 995 1	164.04
60	0.001 008 4	247.68	267.85	0.8206	0.001 004 2	246.06
80	0.001 019 9	330.40	350.80	1.0624	0.001 015 6	328.30
100	0.001 033 7	413.39	434.06	1.2917	0.001 029 0	410.78
120	0.001 049 6	496.76	517.76	1.5102	0.001 044 5	493.59
140	0.001 067 8	586.69	602.04	1.7193	0.001 062 1	576.88
160	0.001 088 5	665.35	687.12	1.9204	0.001 082 1	660.82
180	0.001 112 0	750.95	773.20	2.1147	0.001 104 7	745.59
200	0.001 138 8	837.7	860.5	2.3031	0.001 130 2	831.4
220	0.001 169 3	925.9	949.3	2.4870	0.001 159 0	918.3
240	0.001 204 6	1016.0	1040.0	2.6674	0.001 192 0	1006.9
260	0.001 246 2	1108.6	1133.5	2.8459	0.001 230 3	1097.4
280	0.001 296 5	1204.7	1230.6	3.0248	0.001 275 5	1190.7
300	0.001 359 6	1306.1	1333.3	3.2071	0.001 330 4	1287.9
320	0.001 443 7	1415.7	1444.6	3.3979	0.001 399 7	1390.7
340	0.001 568 4	1539.7	1571.0	3.6075	0.001 492 0	1501.7
360	0.001 822 6	1702.8	1739.3	3.8772	0.001 626 5	1626.6
380					0.001 869 1	1781.4

$\hat{V} [= \text{m}^3/\text{kg}]$; $\hat{U}, \hat{H} [= \text{J/g} = \text{kJ/kg K}]$; $\hat{S} [= \text{J/K} \cdot \text{kg}]$

注意：背面有試題



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TABLE A.6 Thermophysical Properties of Saturated Water^a

Tempera-ture, <i>T</i> (K)	Pressure, <i>p</i> (bars) ^b	Specific Volume (m ³ /kg)		Heat of Vapor- ization, <i>h_f</i> (kJ/kg)	Specific Heat (kJ/kg · K)		Viscosity (N · s/m ²)	Thermal Conductivity (W/m · K)		Prandtl Number	Surface Tension, $\sigma \cdot 10^3$ (N/m)	Expansion Coef- ficient, $\beta \cdot 10^6$ (K ⁻¹)	Temper- ature, <i>T</i> (K)		
		<i>v</i> · 10 ³	<i>v_t</i>		<i>c_p</i>	<i>c_{p,t}</i>		<i>k</i> · 10 ³	<i>k_t</i> · 10 ³						
370	0.9040	1.041	1.861	2265	4.214	2.017	289	11.89	679	24.5	1.80	0.978	59.5	728.7	370
373.15	1.0133	1.044	1.679	2257	4.217	2.029	279	12.02	680	24.8	1.76	0.984	58.9	750.1	373.15
375	1.0815	1.045	1.574	2252	4.220	2.036	274	12.09	681	24.9	1.70	0.987	58.6	761	375
380	1.2869	1.049	1.337	2239	4.226	2.057	260	12.29	683	25.4	1.61	0.999	57.6	788	380
385	1.5233	1.053	1.142	2225	4.232	2.080	248	12.49	685	25.8	1.53	1.004	56.6	814	385
390	1.794	1.058	0.980	2212	4.239	2.104	237	12.69	686	26.3	1.47	1.013	55.6	841	390
400	2.455	1.067	0.731	2183	4.256	2.158	217	13.05	688	27.2	1.34	1.033	53.6	896	400
410	3.302	1.077	0.553	2153	4.278	2.221	200	13.42	688	28.2	1.24	1.054	51.5	952	410
420	4.370	1.088	0.425	2123	4.302	2.291	185	13.79	688	29.8	1.16	1.075	49.4	1010	420
430	5.699	1.099	0.331	2091	4.331	2.369	173	14.14	685	30.4	1.09	1.10	47.2	—	430
440	7.333	1.110	0.261	2059	4.36	2.46	162	14.50	682	31.7	1.04	1.12	45.1	—	440
450	9.319	1.123	0.208	2024	4.40	2.56	152	14.85	678	33.1	0.99	1.14	42.9	—	450
460	11.71	1.137	0.167	1989	4.44	2.68	143	15.19	673	34.6	0.95	1.17	40.7	—	460
470	14.55	1.152	0.136	1951	4.48	2.79	136	15.54	667	36.3	0.92	1.20	38.5	—	470
480	17.90	1.167	0.111	1912	4.53	2.94	129	15.88	660	38.1	0.89	1.23	36.2	—	480
490	21.83	1.184	0.0922	1870	4.59	3.10	124	16.23	651	40.1	0.87	1.25	33.9	—	490
500	26.40	1.203	0.0766	1825	4.66	3.27	118	16.59	642	42.3	0.86	1.28	31.6	—	500
510	31.66	1.222	0.0631	1779	4.74	3.47	113	16.95	631	44.7	0.85	1.31	29.3	—	510
520	37.70	1.244	0.0525	1730	4.84	3.70	108	17.33	621	47.5	0.84	1.35	26.9	—	520
530	44.58	1.268	0.0445	1679	4.95	3.96	104	17.72	608	50.6	0.85	1.39	24.5	—	530

Saturated Steam: Pressure Table (Continued)

Press. (kPa)	Temp. (°C)	Specific Volume		Internal Energy			Enthalpy			Entropy		
		Sat. Liquid	Sat. Vapor	Sat. Liquid	Evap.	Sat. Vapor	Sat. Liquid	Evap.	Vapor	Sat. Liquid	Evap.	Sat. Vapor
<i>P</i>	<i>T</i>	\hat{V}^L	\hat{V}^V	\hat{U}^L	$\Delta \hat{U}$	\hat{U}^V	\hat{H}^L	$\Delta \hat{H}$	\hat{H}^V	\hat{S}^L	$\Delta \hat{S}$	\hat{S}^V
1.00	179.91	0.001 127	0.194 44	761.68	1822.0	2583.6	762.81	2015.3	2778.1	2.1387	4.4478	6.5865
1.10	184.09	0.001 133	0.177 53	780.09	1806.3	2586.4	781.34	2000.4	2781.7	2.1792	4.3744	6.5536
1.20	187.99	0.001 139	0.163 33	797.29	1791.5	2588.8	798.65	1986.2	2784.8	2.2166	4.3067	6.5233
1.30	191.64	0.001 144	0.151 25	813.44	1777.5	2591.0	814.93	1972.7	2787.6	2.2515	4.2438	6.4953
1.40	195.07	0.001 149	0.140 84	828.70	1764.1	2592.8	830.30	1959.7	2790.6	2.2842	4.1850	6.4693
1.50	198.32	0.001 154	0.131 77	843.16	1751.3	2594.5	844.89	1947.3	2792.2	2.3150	4.1298	6.4448
1.75	205.76	0.001 166	0.113 49	876.46	1721.4	2597.8	878.50	1917.9	2796.4	2.3851	4.0044	6.3896
2.00	212.42	0.001 177	0.099 63	906.44	1693.8	2600.3	908.79	1890.7	2799.5	2.4474	3.8935	6.3409
2.25	218.45	0.001 187	0.088 75	933.83	1668.2	2602.0	936.49	1865.2	2801.7	2.5035	3.7937	6.2972
2.5	223.99	0.001 197	0.079 98	959.11	1644.0	2603.1	962.11	1841.0	2803.1	2.5547	3.7028	6.2575
3.00	233.90	0.001 217	0.066 68	1004.78	1599.3	2604.1	1008.42	1795.7	2804.2	2.6457	3.5412	6.1869
3.5	242.60	0.001 235	0.057 07	1045.43	1558.3	2603.7	1049.75	1753.7	2803.4	2.7253	3.4000	6.1253
4	250.40	0.001 252	0.049 78	1082.31	1520.0	2602.3	1087.31	1714.1	2801.4	2.7964	3.2737	6.0701
5	263.99	0.001 286	0.039 44	1147.81	1449.3	2597.1	1154.23	1640.1	2794.3	2.9202	3.0532	5.9734
6	275.64	0.001 319	0.032 44	1205.44	1384.3	2589.7	1213.35	1571.0	2784.3	3.0267	2.8625	5.8892
7	285.88	0.001 351	0.027 37	1257.55	1323.0	2580.5	1267.00	1505.1	2772.1	3.1211	2.6922	5.8133
8	295.06	0.001 384	0.023 52	1305.57	1264.2	2569.8	1316.64	1441.3	2758.0	3.2068	2.5364	5.7432
9	303.40	0.001 418	0.020 48	1350.51	1207.3	2557.8	1363.26	1378.9	2742.1	3.2858	2.3915	5.6772
10	311.06	0.001 452	0.018 026	1393.04	1151.4	2544.4	1407.56	1317.1	2724.7	3.3596	2.2544	5.6141
11	318.15	0.001 489	0.015 987	1433.7	1096.0	2529.8	1450.1	1255.5	2705.6	3.4295	2.1233	5.5527
12	324.75	0.001 527	0.014 263	1473.0	1040.7	2513.7	1491.3	1193.6	2684.9	3.4962	1.9962	5.4924
13	330.93	0.001 567	0.012 780	1511.1	985.0	2496.1	1531.5	1130.7	2662.2	3.5606	1.8718	5.4323
14	336.75	0.001 611	0.011 485	1548.6	928.2	2476.8	1571.1	1066.5	2637.6	3.6232	1.7485	5.3717
15	342.24	0.001 658	0.010 337	1585.6	869.8	2455.5	1610.5	1000.0	2610.5	3.6848	1.6249	5.3098
16	347.44	0.001 711	0.009 306	1622.7	809.0	2431.7	1650.1	930.6	2580.6	3.7461	1.4994	5.2455
17	352.37	0.001 770	0.008 364	1660.2	744.8	2405.0	1690.3	856.9	2547.2	3.8079	1.3698	5.1777
18	357.06	0.001 840	0.007 489	1698.9	675.4	2374.3	1732.0	777.1	2509.1	3.8715	1.2329	5.1044
19	361.54	0.001 924	0.006 657	1739.9	598.1	2338.1	1776.5	688.0	2464.5	3.9388	1.0839	5.0228
20	365.81	0.002 036	0.005 834	1785.6	507.5	2293.0	1826.3	583.4	2409.7	4.0139	0.9130	4.9269
21	369.89	0.002 207	0.004 952	1842.1	388.5	2230.6	1888.4	446.2	2334.6	4.1075	0.6938	4.8013
22	373.80	0.002 742	0.003 568	1961.9	125.2	2087.1	2022.2	143.4	2165.6	4.3110	0.2216	4.5327
22.09	374.14	0.003 155	0.003 155	2029.6	0.0	2029.6	2099.3	0.0	2099.3	4.4298	0.0	4.4298

\hat{V} [=] m³/kg; \hat{U}, \hat{H} [=] J/g = kJ/kg; \hat{S} [=] kJ/kg K