## 國立中央大學101學年度碩士班考試入學試題卷

所別:環境工程研究所碩士班 甲組(一般生) 科目:衛生工程 共 / 頁 第 / 頁 本科考試禁用計算器 \*請在試卷答案卷(卡)內作答

- 1. Explain the followings: (25%)
  - (a) Disinfection by-products (DBPs)
  - (b) Endocrine disruptor
  - (c) Sweep coagulation
  - (d) Endogenous decay
  - (e) Overflow rate
- 2. Alum  $[Al_2(SO_4)_3 \cdot 18 H_2O]$  is the mostly used coagulant. One problem of using alum is that the consumption of alkalinity. (20%)
  - (a) Complete and balance the precipitation of alumina upon the addition of alum.
  - (b) Form (a), calculate the required alkalinity per mg/L of alum is added.
  - (c) What are other possible problems of using alum as coagulant?
- 3. Compare the advantages and disadvantages of biofilm (e.g., trickling filter) and activated sludge systems. (15%)
- 4. Compare the water quality between groundwater and river water and discuss the differences in the potable water treatment processes when using them as source waters. (15%)
- Analysis of a groundwater sample gives the following: [Ca<sup>2+</sup>] = 60 mg/L, [Mg<sup>2+</sup>] = 24 mg/L, [HCO<sub>3</sub><sup>-</sup>] = 200 mg/L as CaCO<sub>3</sub>. Calculate the dosages of lime (Ca(OH)<sub>2</sub>) and soda ash (Na<sub>2</sub>CO<sub>3</sub>) (10%) (Note: only [HCO<sub>3</sub><sup>-</sup>] is presented in mg/L as CaCO<sub>3</sub>)(M.W.: Ca = 40, Mg = 24, Na = 23, O = 16, C = 12).(15%)
- 6. From the view of sustainable water treatments, mineralization of organic compounds (e.g., BOD, NBOD, and etc) in activated sludge system is considered energy-wasted for some water-reclamation purposes. Why is that? What could be more sustainable ways? (10%)